

Spectrophotometric Nitrate Measurement in Natural Waters

CONDESA Group 1

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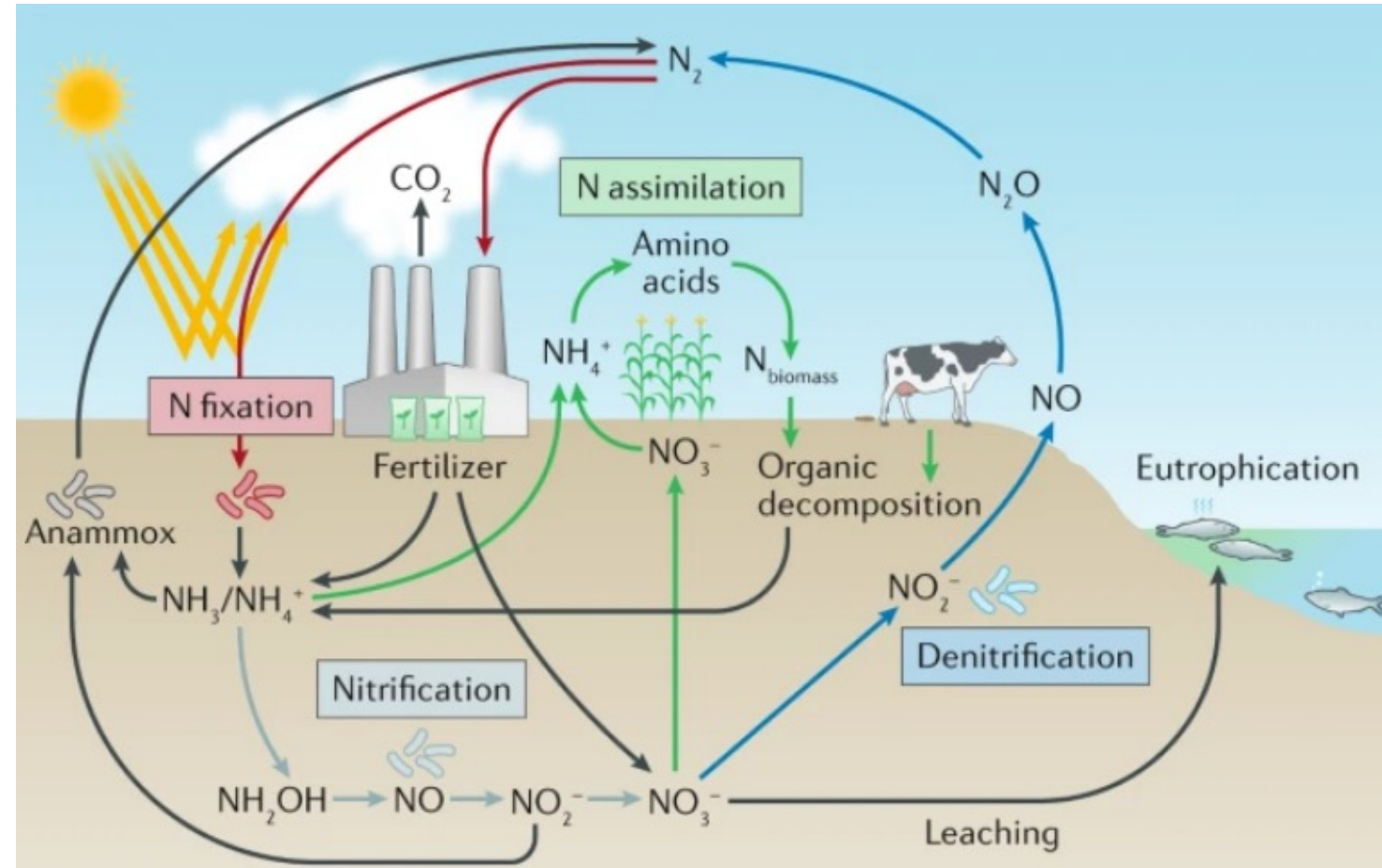
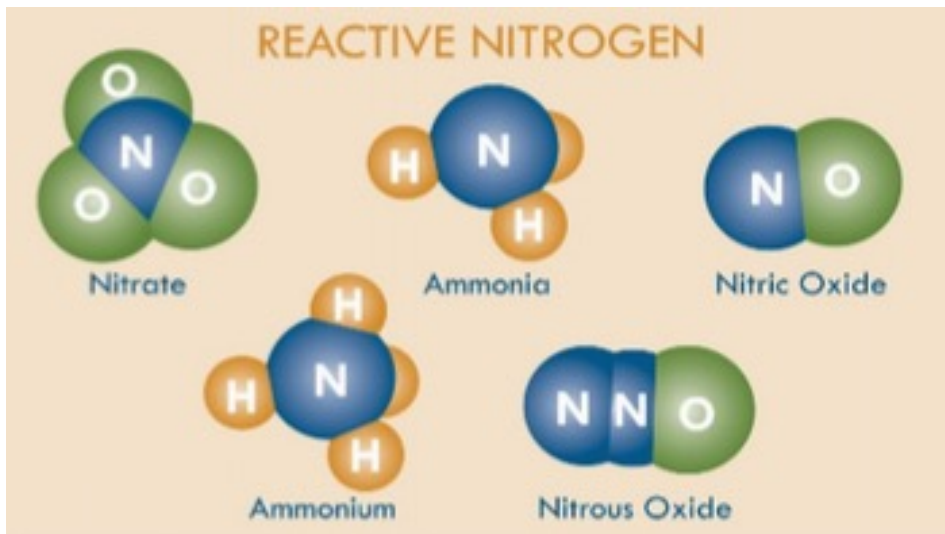


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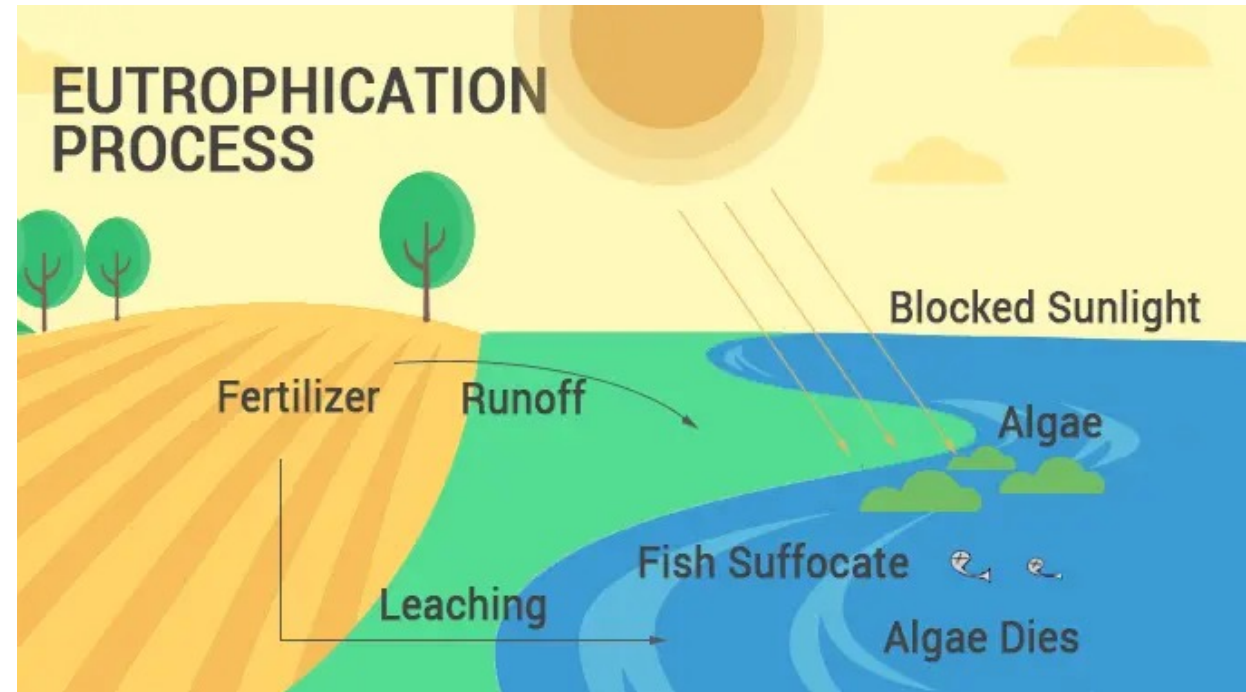
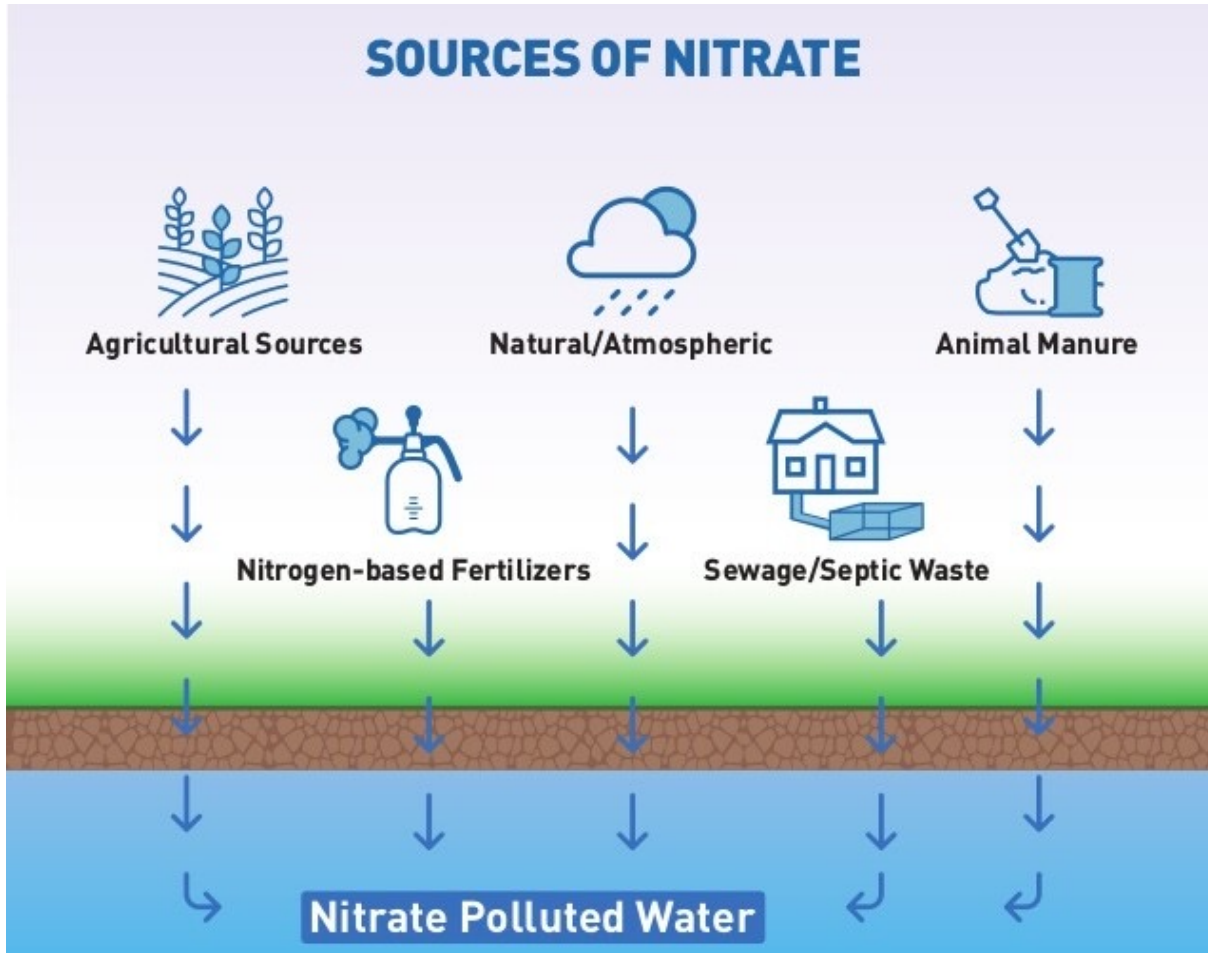
Nitrogen Background





Nitrogen Background

SOURCES OF NITRATE

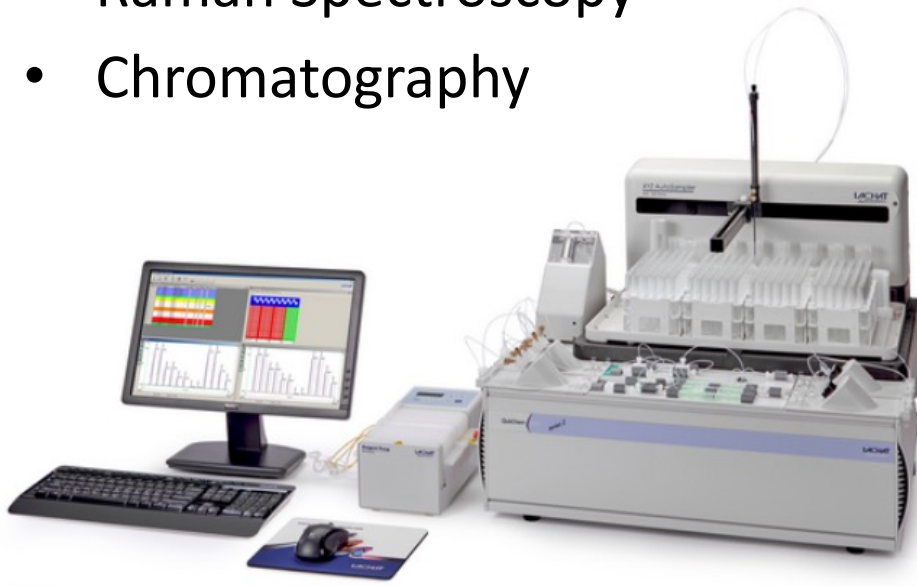




Methods Background

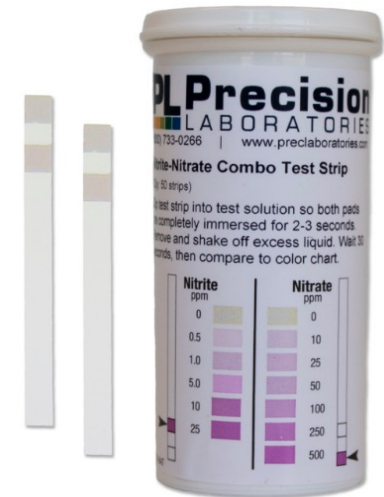
Lab Methods

- Flow Injection Colorimetry
- Fluorescence
- Raman Spectroscopy
- Chromatography



Field Methods

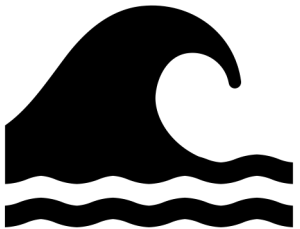
- Nitrate Test Strips
- Field Probes





Project Aims

Develop a low cost, portable, and easy to use sensor capable of detecting nitrate in natural waters from 0.2 ppm N – 40 ppm N



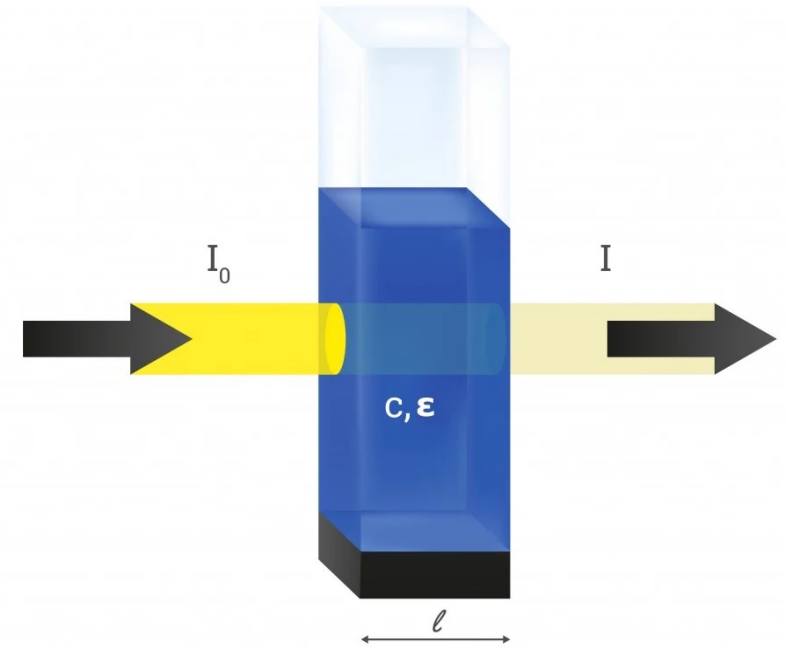
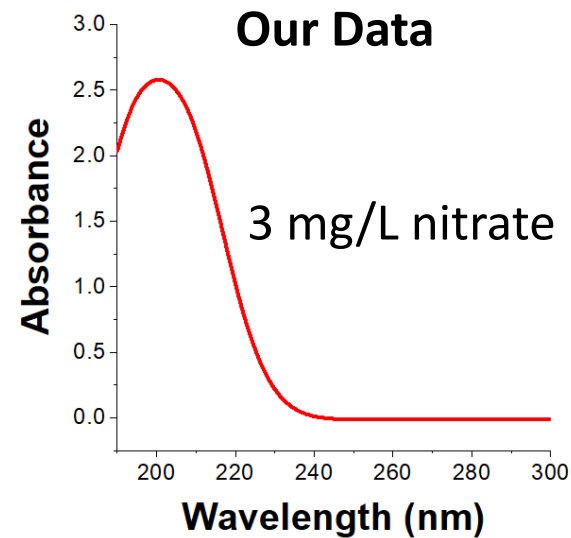
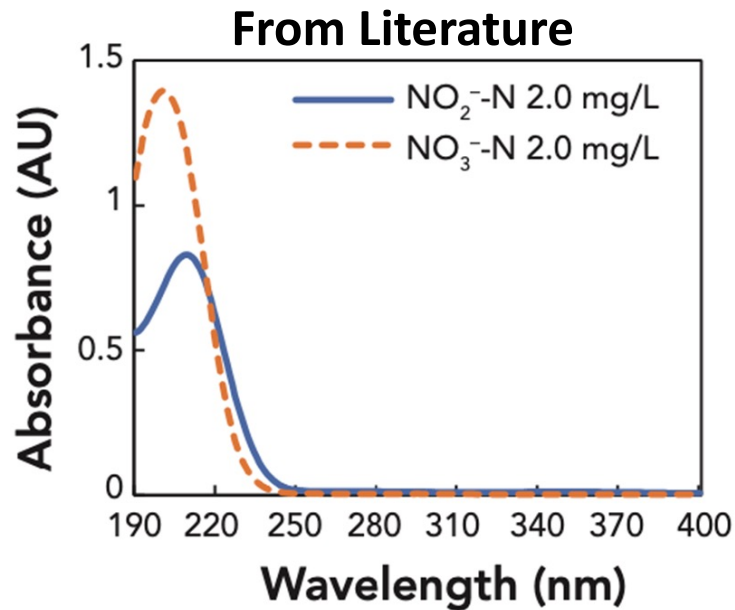


UV Absorption by Nitrate Ions

Beer-Lambert's Law

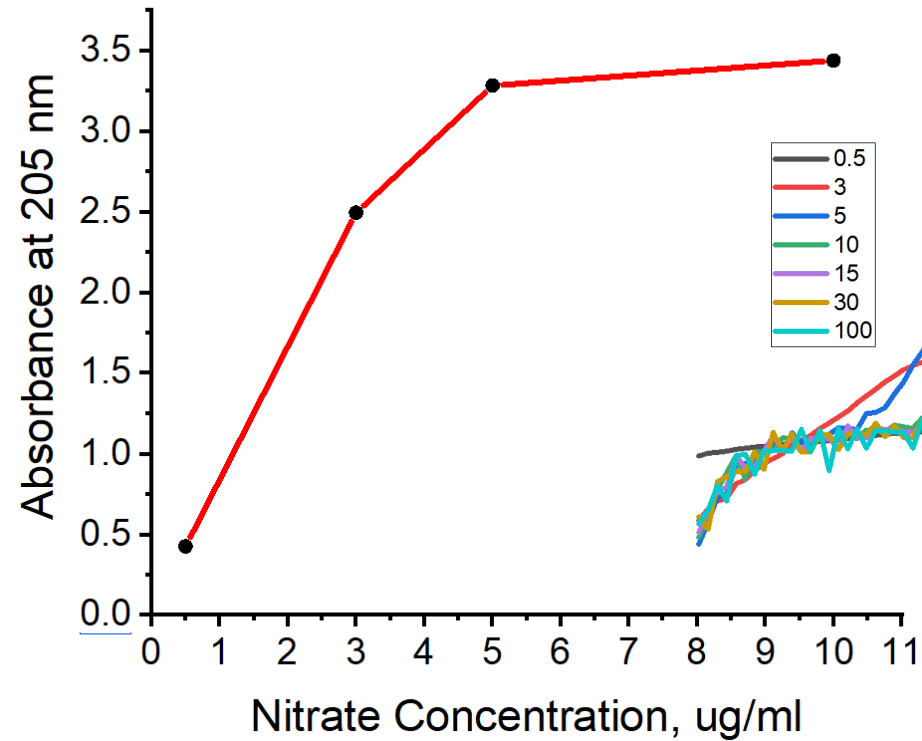
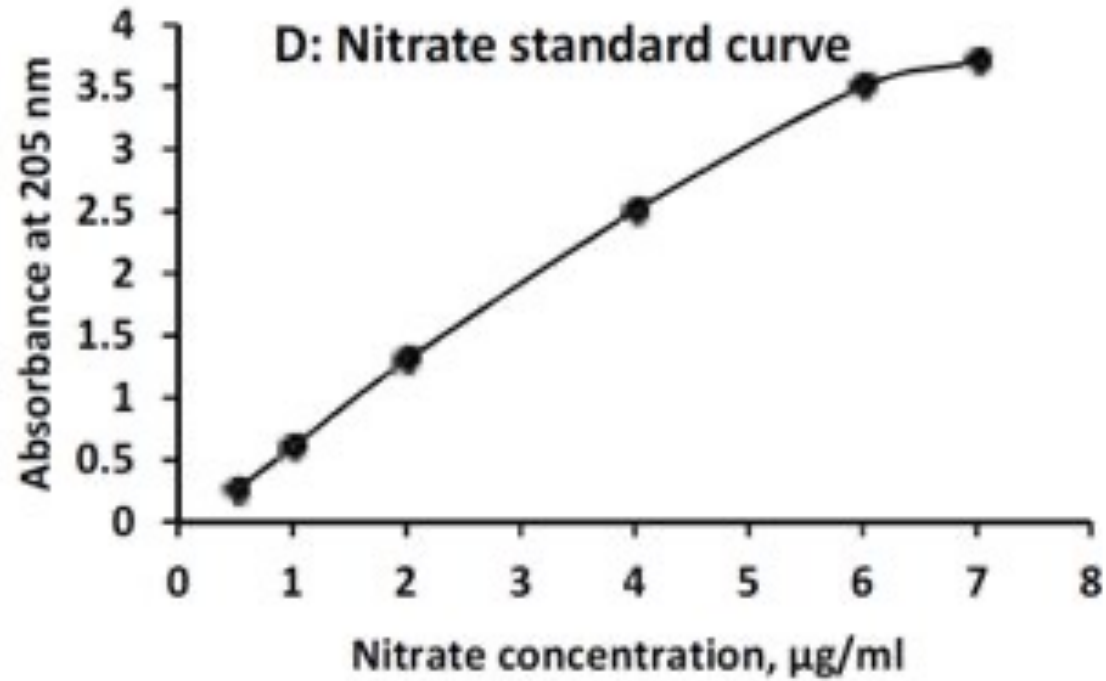
$$A = \log_{10} \left(\frac{I_0}{I} \right) = \epsilon c l$$

A = absorbance
I = intensity
 ϵ = molar absorptivity
L = path length
c = concentration of analyte





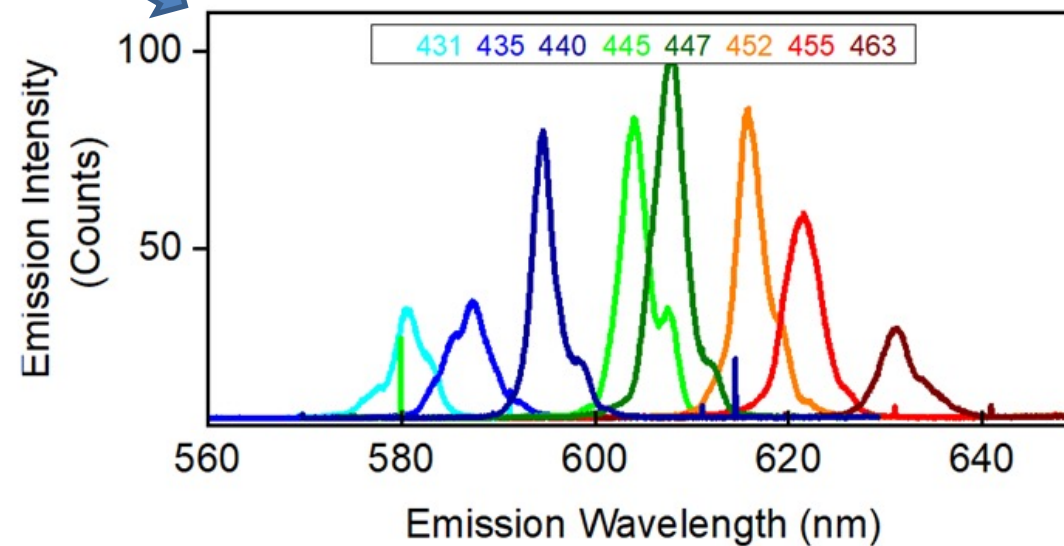
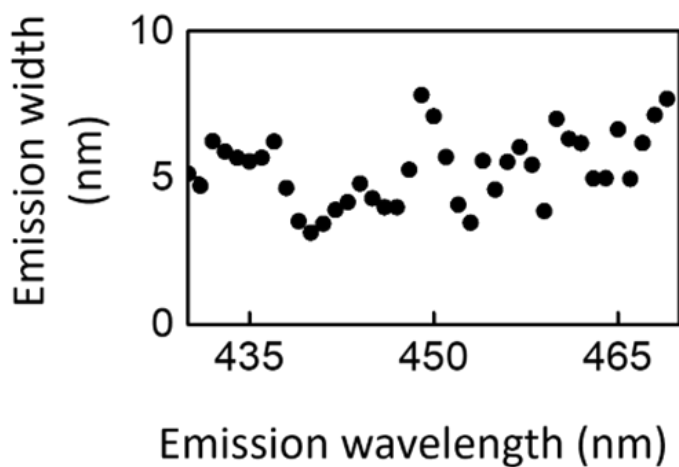
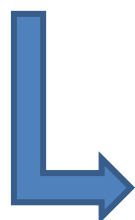
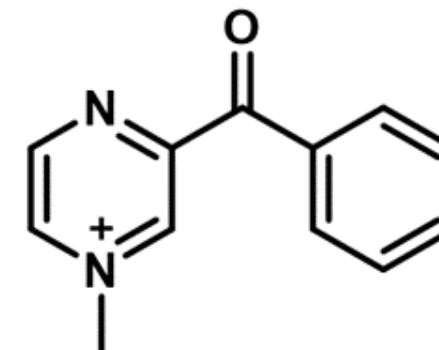
Limitations of Beer-Lambert's Law





Organic Luminescent Molecules

- Substituted Benzoyl Pyraziniums [synthesized in Baxter Lab] show interesting optical properties in solution and solid state
- Solid state studies show emission **tunability** with excitation wavelength
- Concentration-dependent **tunable** emission
- Significantly narrowed emission, with a FWHM of 4 nm





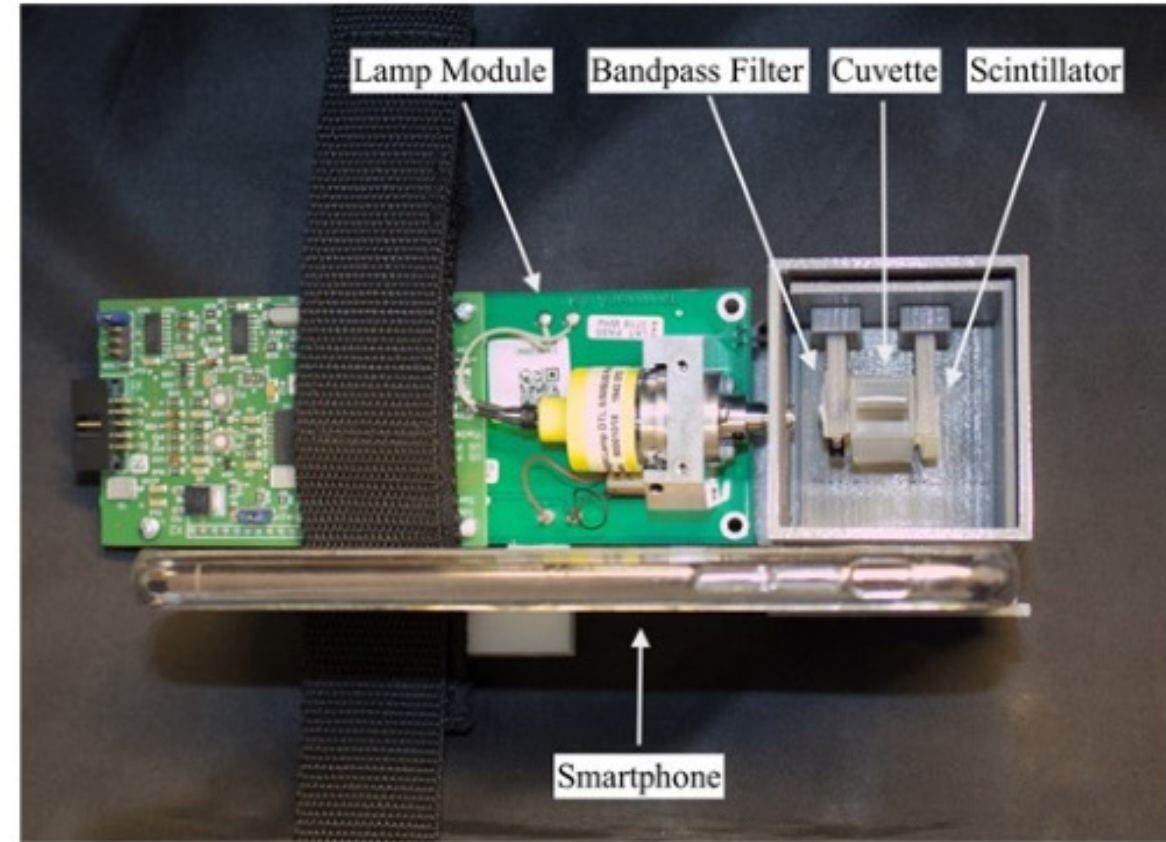
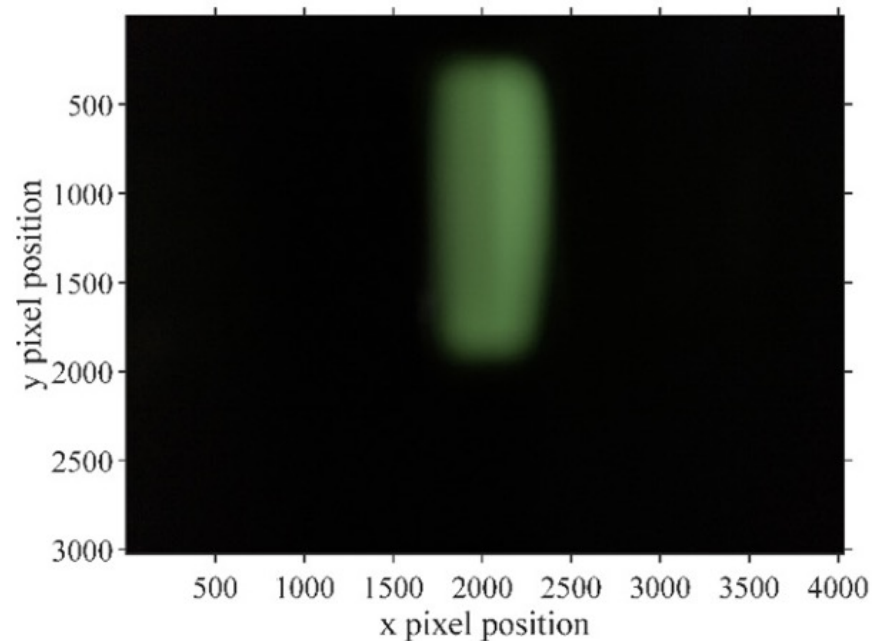
Prototype Spectrophotometer

Water quality assessment using a portable UV optical absorbance nitrate sensor with a scintillator and smartphone camera

JMDFP Ingles¹, TM Louw² and MJ Booysen¹

¹Department of E&E Engineering, Stellenbosch University, Stellenbosch, South Africa

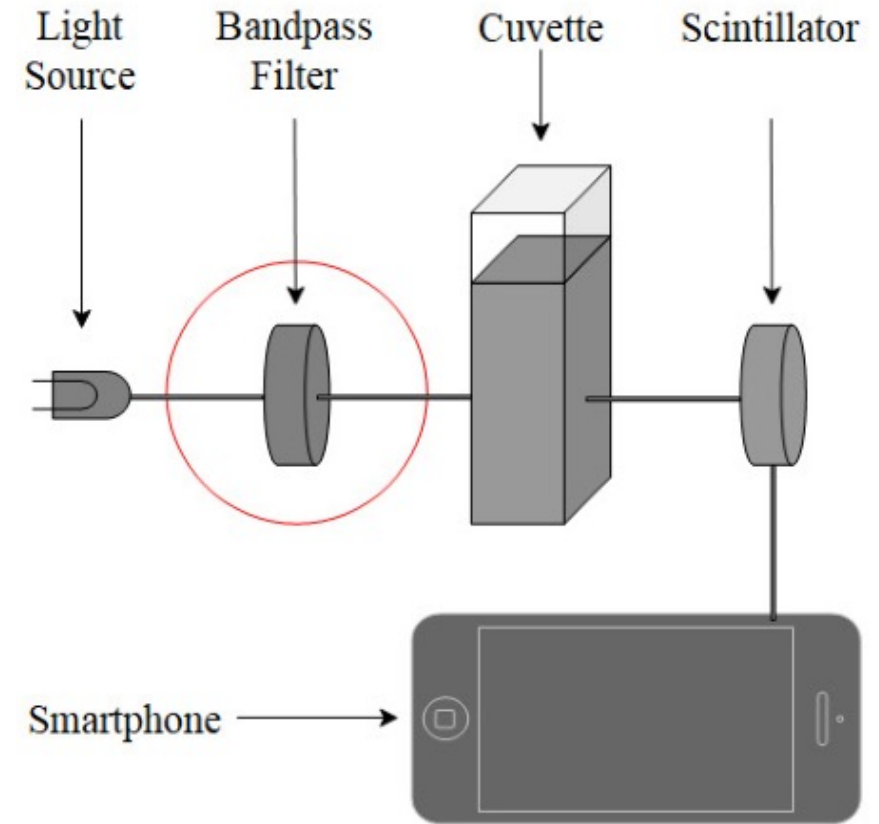
²Department of Process Engineering, Stellenbosch University, Stellenbosch, South Africa





Proposed Optical Sensor

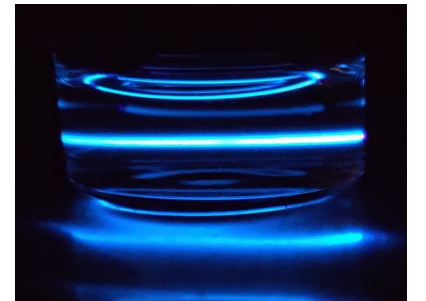
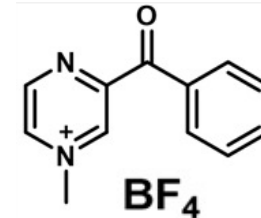
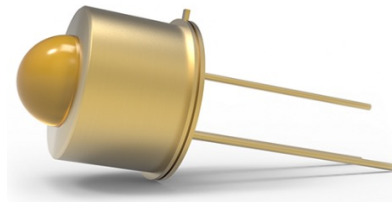
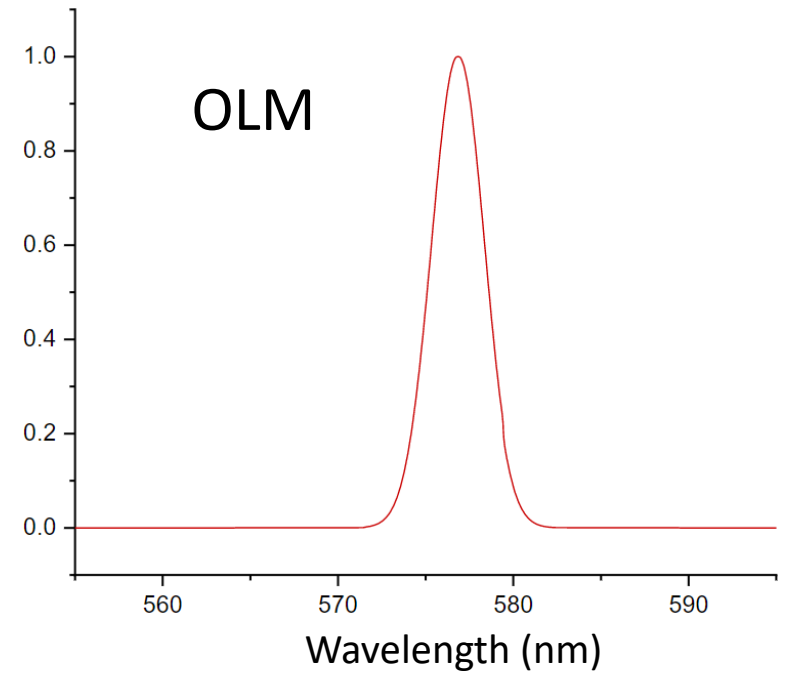
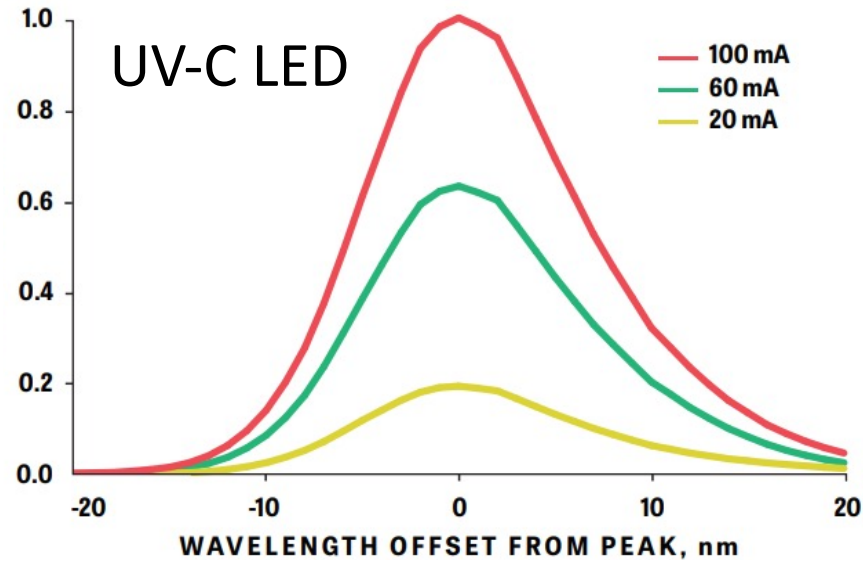
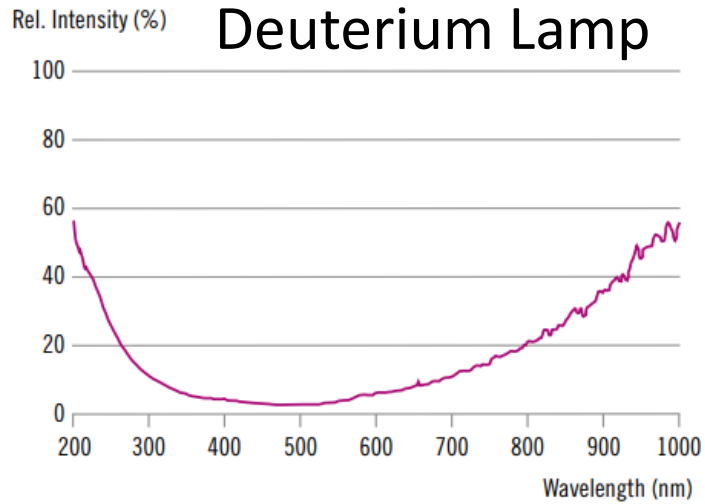
Original	Improvements
Deuterium lamp \$525	UV-C LED < \$300 Lower power, longer lasting
Quartz cuvette \$75	\$15 on Amazon
200 nm bandpass \$75	Possibly not needed
Green glass scintillator \$75	No improvements found
Total: \$750	Total: \$300-\$500





Comparing Light Sources

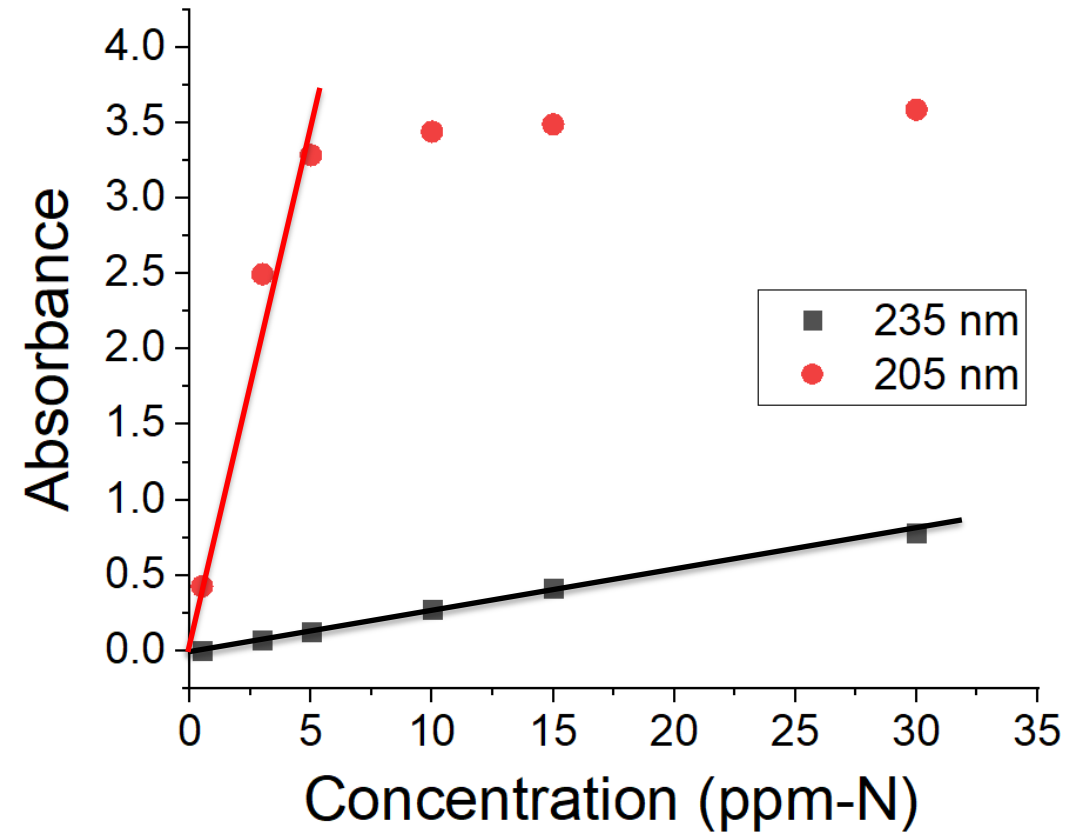
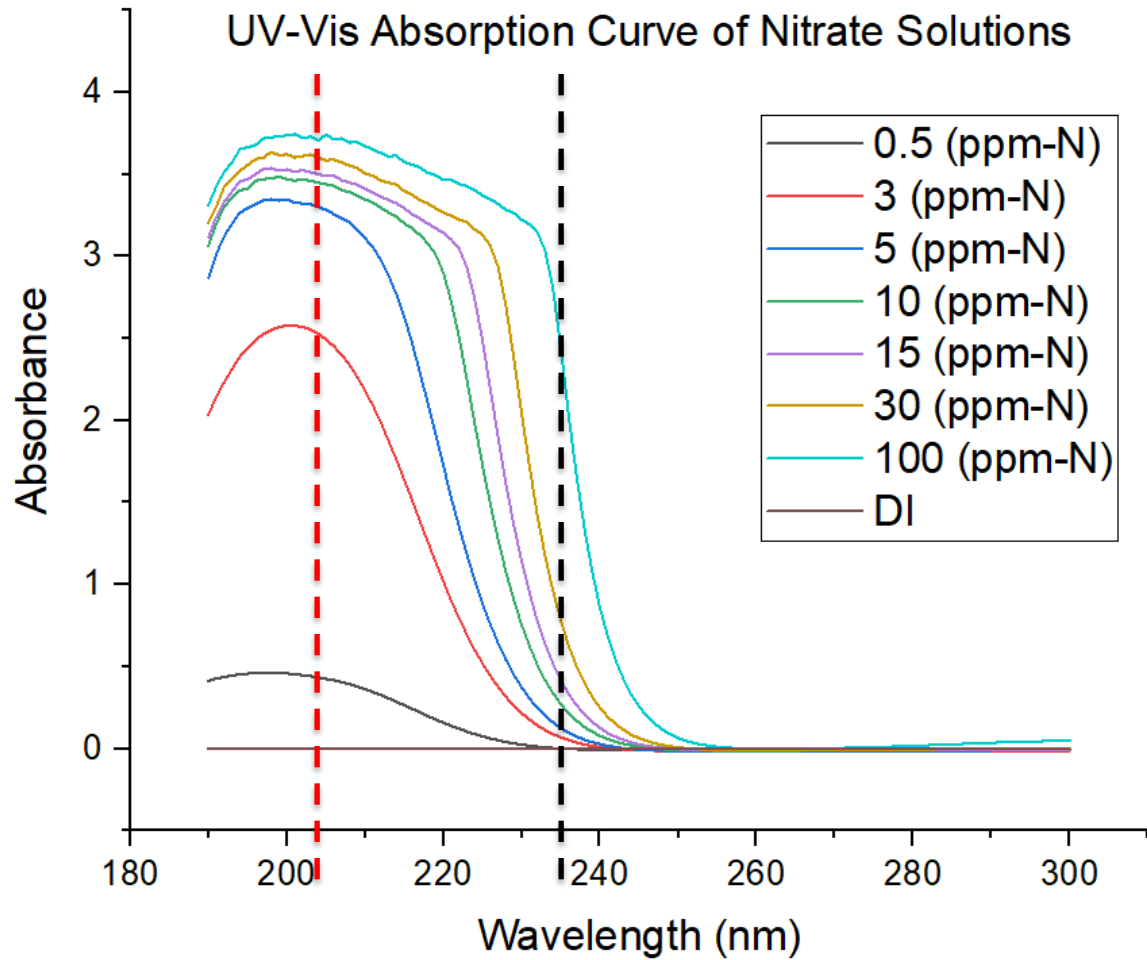
Spectral Distribution



LEDs and OLMs = low power consumption and narrow spectral emission

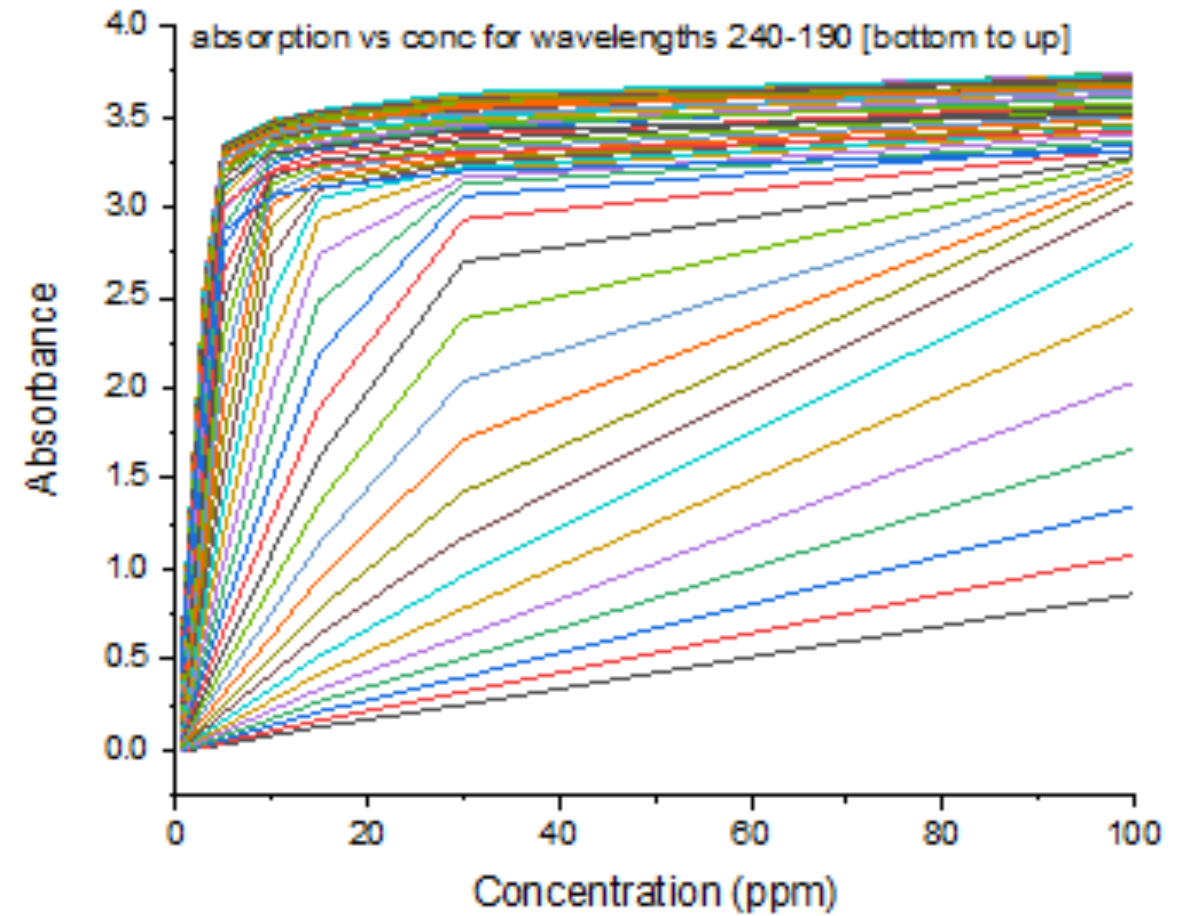
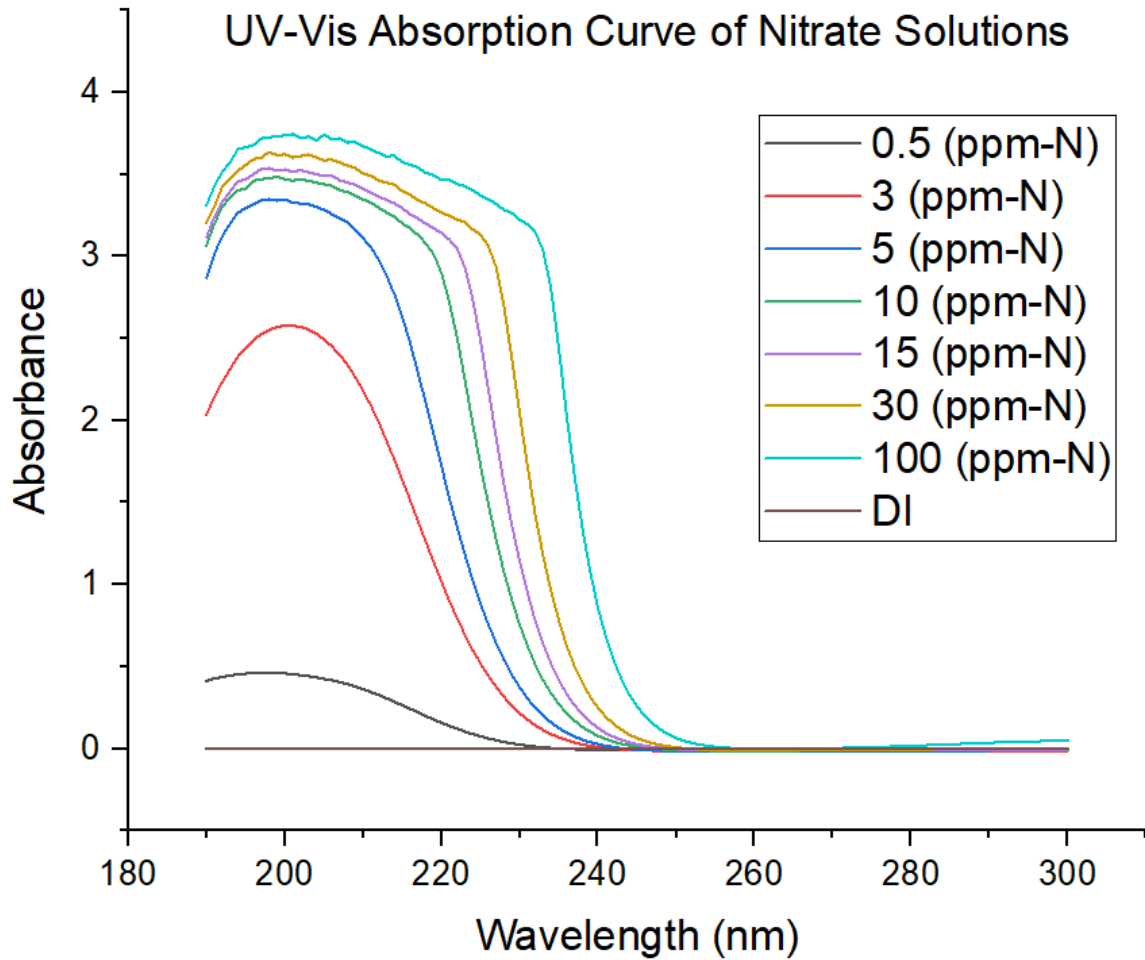


Nitrate Absorption Test



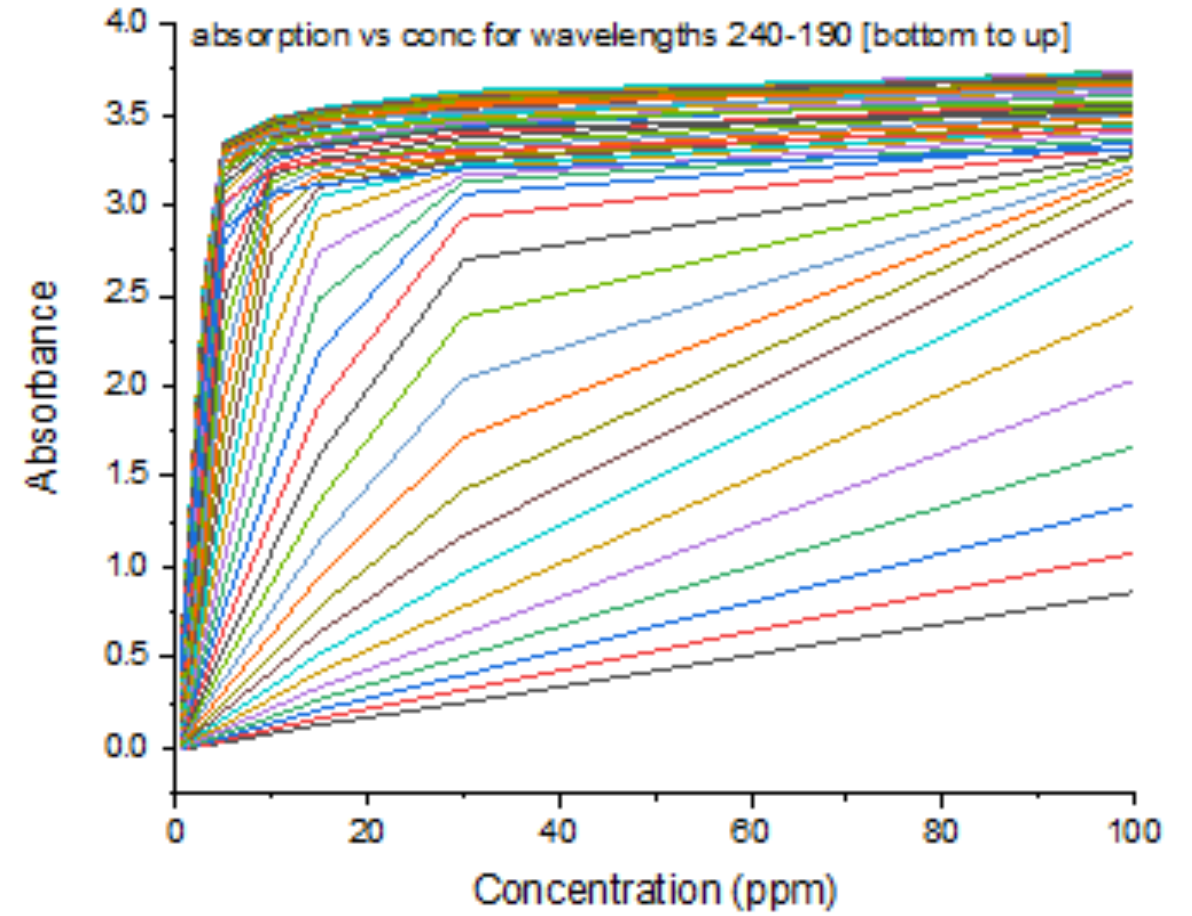
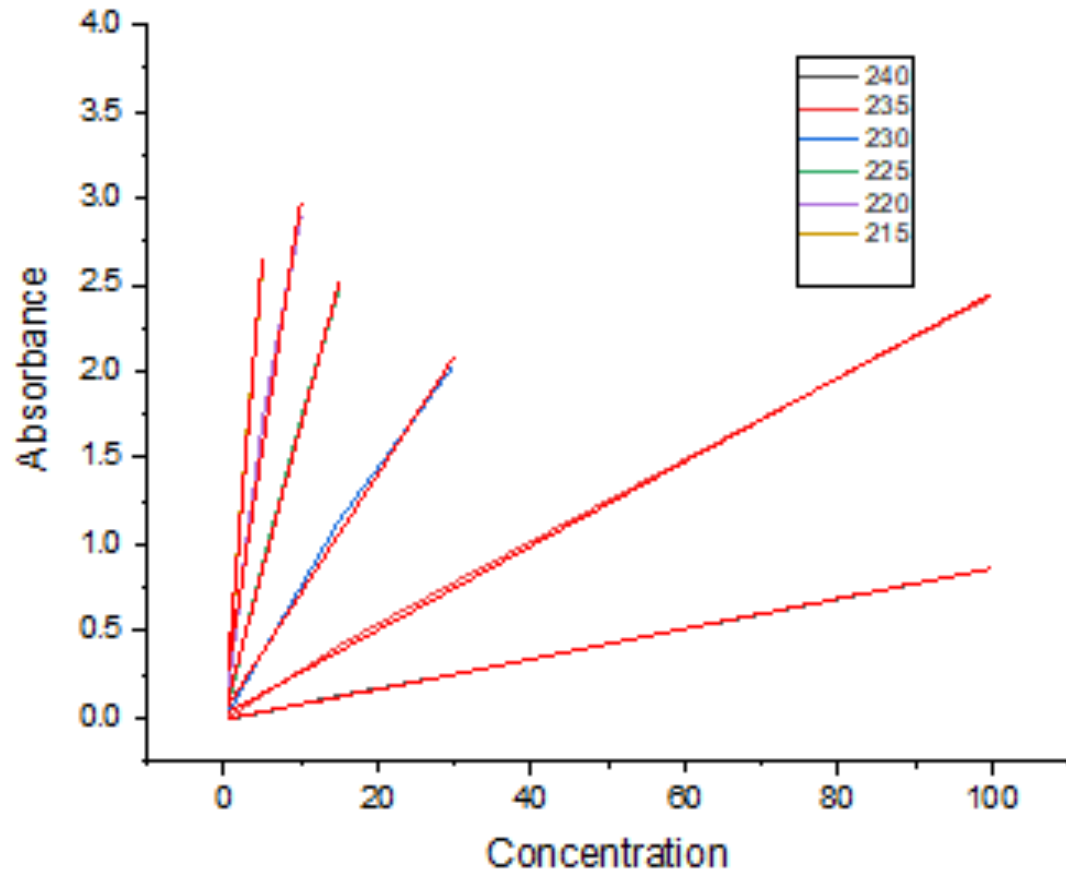


Determining Concentration



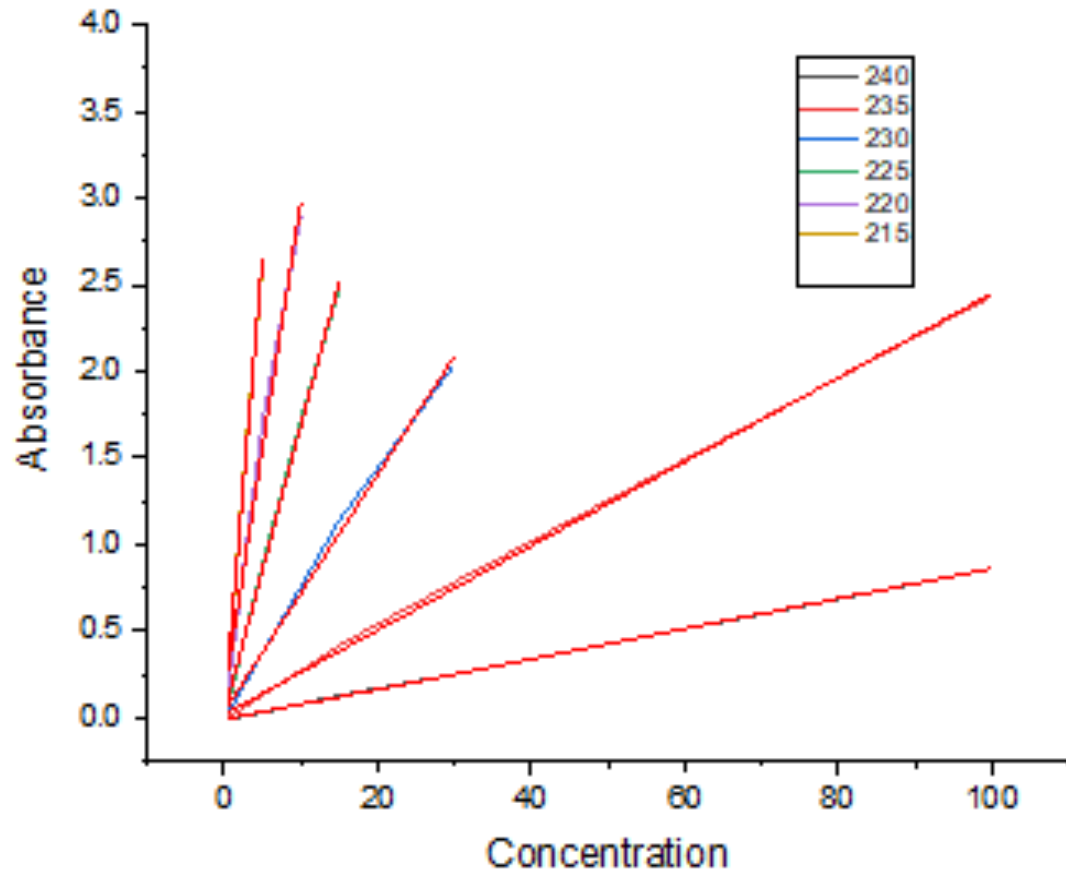


Determining Concentration





Determining Concentration



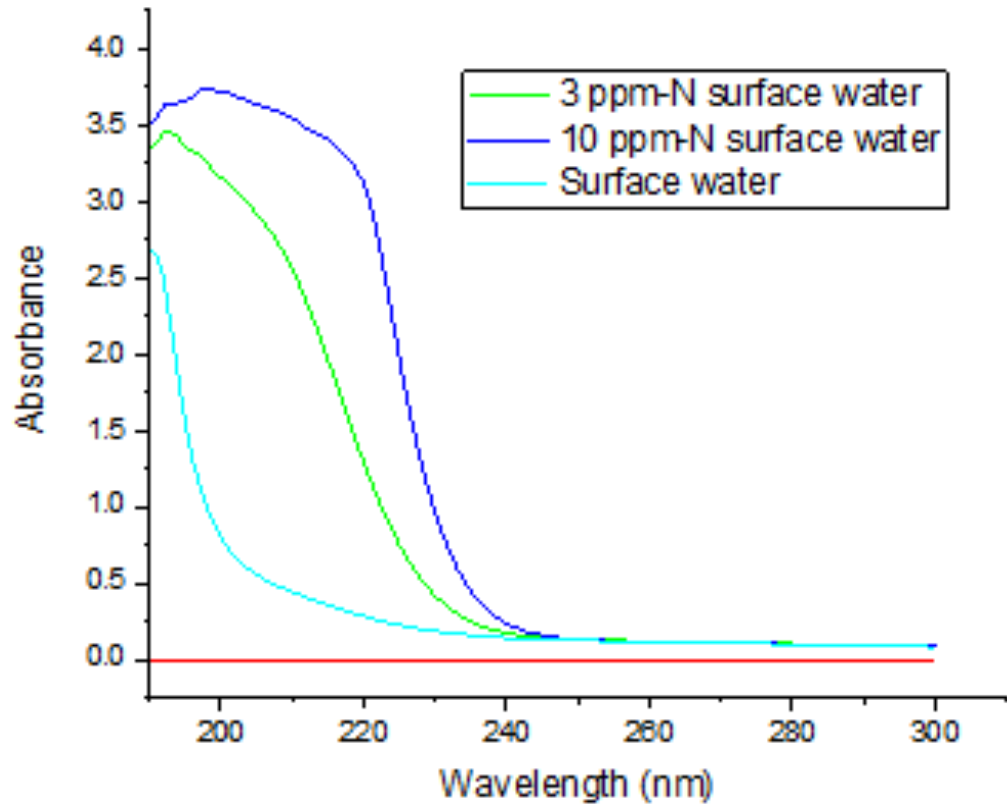
wavelength	slope
240	0.00873
235	0.02432
230	0.06849
225	0.16638
220	0.28577
215	0.52674

Wavelength	predicted concentration
215	0.503
215	3.111
215	4.995
220	10.145
225	14.937
230	29.793
235	100.156

$$A = \epsilon C$$



Determining Concentration



wavelength	slope
240	0.00873
235	0.02432
230	0.06849
225	0.16638
220	0.28577
215	0.52674



Determining Concentration

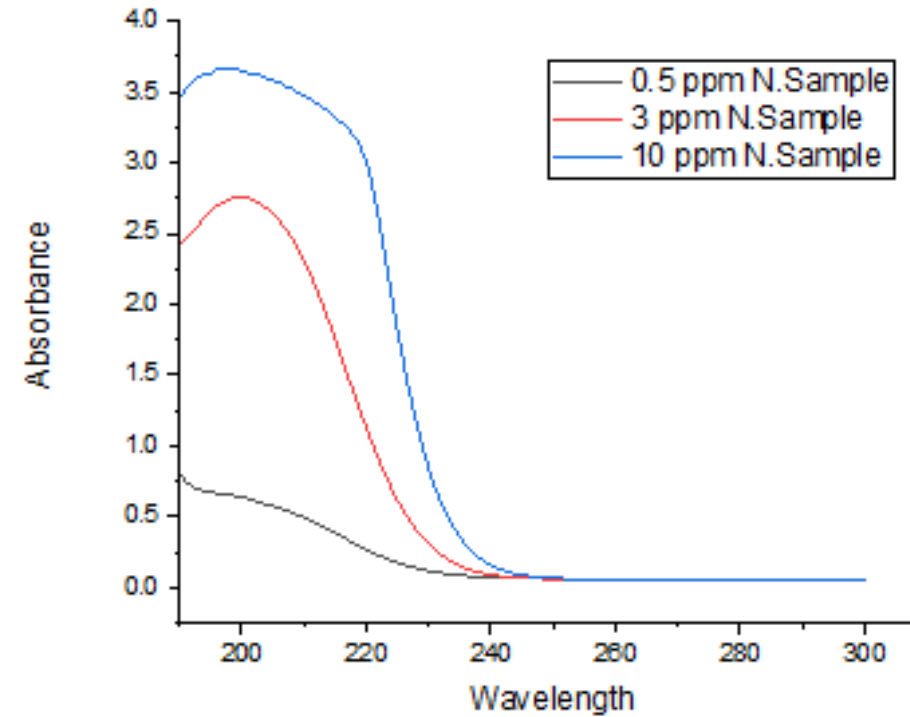
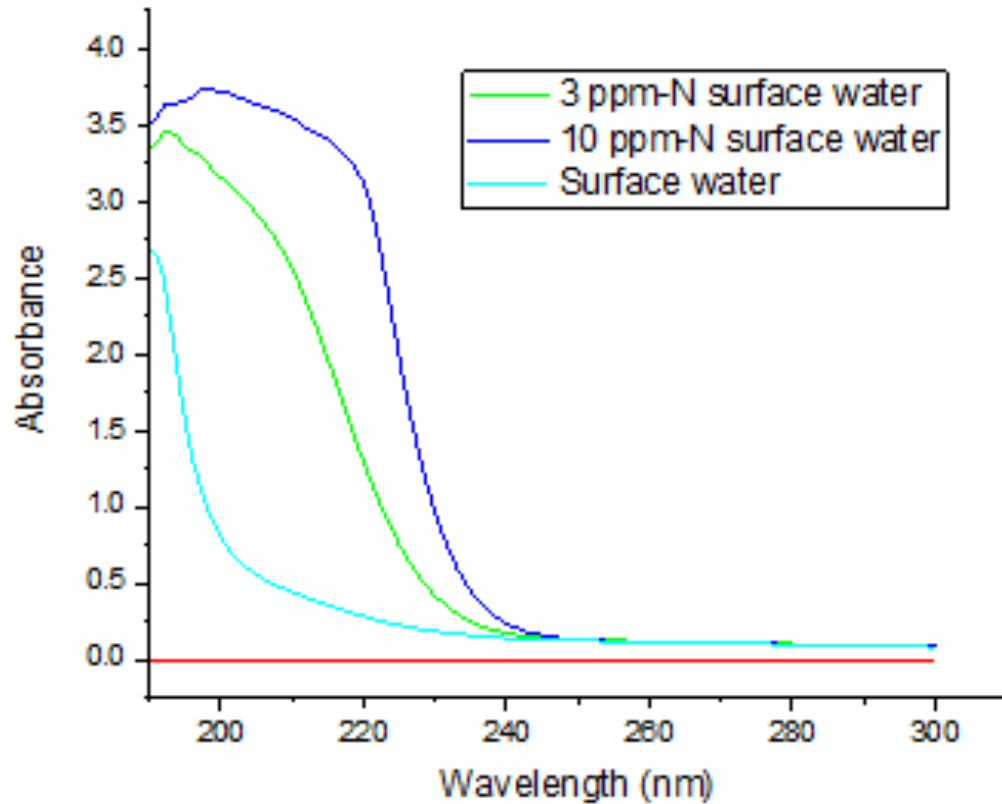
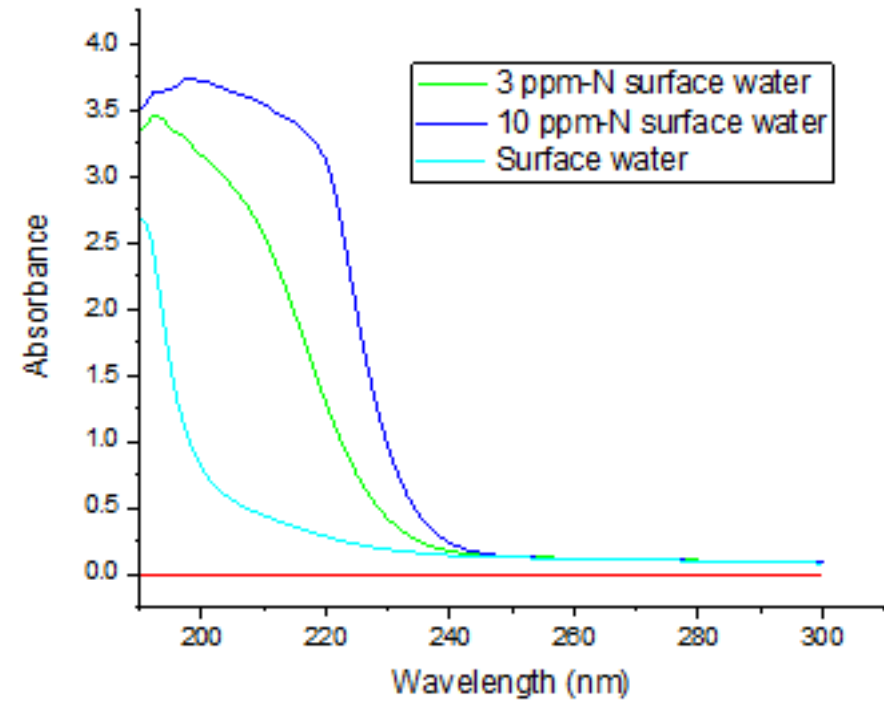
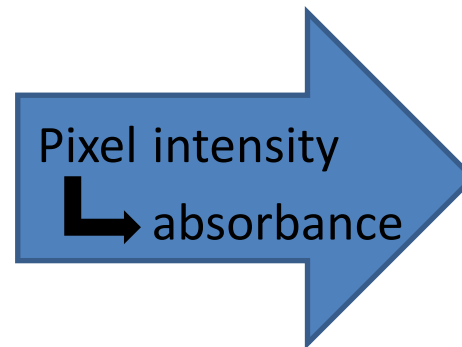
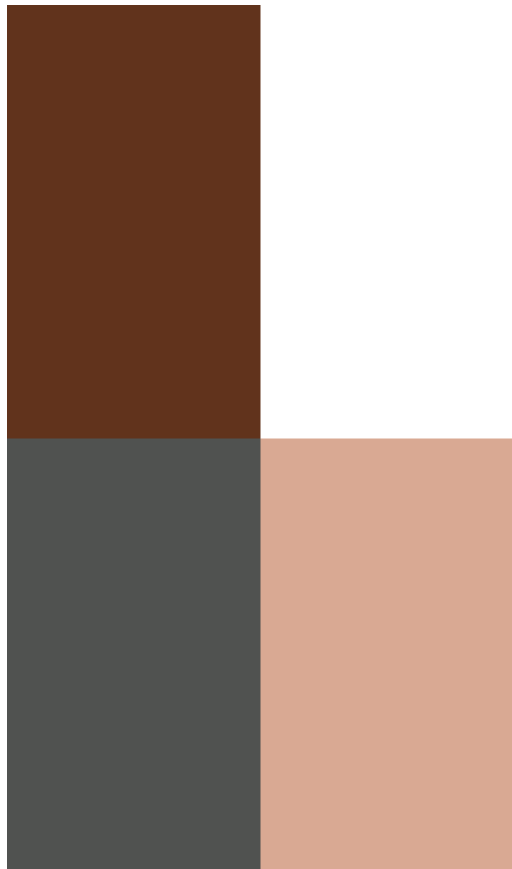




Image Analysis

Deconvolution algorithms



Other solutions: utilize a photodiode to capture light signal from sample



Interface

Nitrate Testing

Tuesday June 21, 2022

How much nitrate is in the water? Let's find out!

0-5 ppm-N 5-10 ppm-N 10-15 ppm-N 15-30 ppm-N 30-50 ppm-N

Concentration Output

```
calculated concentration @ 205.0:  
sample 1: 0.6344831456088803 ± 0.3013497413212077  
sample 2: 5.427491429970619 ± 0.0352283065333071  
sample 3: 4.31862400016324 ± 0.044273669528691315  
calculated concentration @ 235.0:  
sample 1: 2.765503833483671 ± 0.00845769675248732  
sample 2: 13.688113660825854 ± 0.0017087666986858499  
sample 3: 6.092397409731739 ± 0.0038391771282162593
```

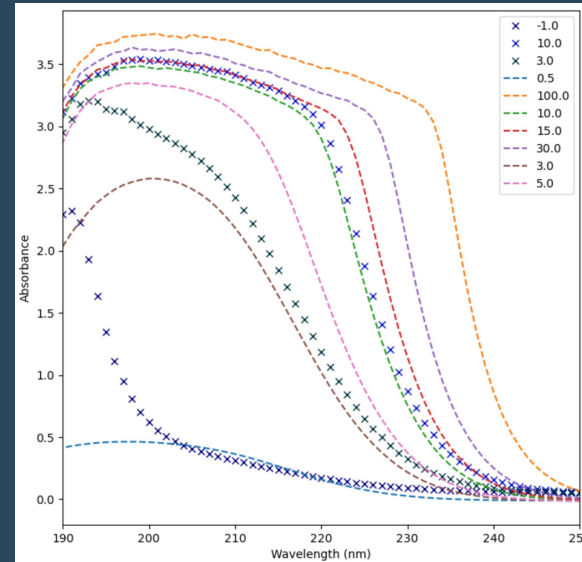
Control Panel

Go

Exit

Click 'Go' to sample more water.

Plotting



Acknowledgements

- Thank you to our group advisors and the CONDESA team
- We are incredibly grateful for the funding and opportunity provided by NRT CONDESA, funded by National Science Foundation grant no. 2125510

